



Learn about elements as you turn a citrus fruit into a thirst-quenching fizzy drink.

### What you need

- Knife
- 1 lemon
- Juicer
- Bowl or jug
- Glass
- Water
- Ice cubes
- 1tsp sugar
- 1tsp bicarbonate



### Instructions

Cut the lemon in **CLEANER** half. Place the juicer over a bowl or jug (if it does not have a juice collector) and squeeze each lemon half over the juicer. Pour the juice into a glass, fill the rest of the glass with water and add ice. Add sugar to make it a little sweeter. Now, for the science. Add a teaspoon of bicarbonate of soda to the juice and watch what happens. You should see lots of small bubbles forming instantly. Congratulations – you've used chemistry

to create your own fizzy lemonade.

# NATURAL

The citric acid in lemon juice is great at breaking down certain types of dirt, making it easier to wipe away grime.

## How does it work?

Lemon juice

contains an acid.

A chemical reaction occurs when two substances are combined to form a new substance. In this experiment, a reaction occurs between an acid and a base. The acid is the lemon juice. The base is the bicarbonate of soda. When they are mixed together, carbon dioxide is released. Carbon dioxide (CO<sub>3</sub>) is a colourless and

odourless gas made from one atom of carbon and two atoms of oxygen joined together. When the gas forms within a liquid, it creates bubbles.

#### **MEET THE ELEMENTS**

One carbon atom and two oxygen atoms join to make carbon dioxide.





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